

Tracking Bismuth Antiulcer Drug Uptake in Single Helicobacter pylori Cells

Cheuk-Nam Tsang,[†] Koon-Sing Ho,[†] Hongzhe Sun,^{*} and Wing-Tat Chan^{*}

Department of Chemistry, The University of Hong Kong, Pokfulam Road, Hong Kong, P. R. China

Supporting Information

ABSTRACT: Bismuth-based drugs have long been used for the treatment of *Helicobacter pylori* infection. In this work, the metal content in *H. pylori* was monitored at the single-cell level by time-resolved inductively coupled plasma mass spectrometry, and ~2.9 × 10⁷ Mg atoms/cell was determined for the wild-type. Bacteria treated with a Bi antiulcer drug deposited nearly 1.0×10^6 Bi atoms/cell, whereas the uptake process took ~3 h to reach the half-maximum. Interference of ferric ions on bismuth uptake was demonstrated, suggesting that the metallodrug can utilize certain iron-transport pathways in the pathogen. The approach provides a general strategy for monitoring metals in single cells, facilitating exploration of metal-relevant bioprocesses.

t has been well-known that at least 25 elements are essential for Llife, among which most are metals or metalloids.¹ Metals are commonly involved in (patho)physiological processes. Malfunction of metal regulatory proteins causes diseases such as Menkes disease and Wilson's disease, which are due to copper deficiency and overload, respectively.² Metal complexes are also used as therapeutic and diagnosis agents.³ The success of the platinum-based anticancer drug cisplatin has stimulated the use of metal-containing compounds in medicine.⁴ Bismuth complexes, such as colloidal bismuth subcitrate (CBS) and ranitidine bismuth citrate (RBC), together with antibiotics, have been used for decades for the treatment of Helicobacter pylori infections and peptic ulcers.^{3a,5} New techniques and strategies such as metallomics and metalloproteomics have been developed for the systematic study of biometals and metallodrugs.^c Inductively coupled plasma mass spectrometry (ICP-MS) is one of the useful methods because of its high sensitivity. For example, ICP-MS was used to determine the zinc content in Escherichia coli to understand the regulation of intracellular Zn.⁷ It has also been used in genome-wide metallomic surveys to search genes and gene networks that are associated with metal hemostasis in Arabidopsis thaliana and Saccharomyces cerevisiae.⁸

ICP-MS measurements on a biological sample typically require acid digestion of the sample, which is tedious and prone to contamination and sample loss. Single-cell analysis using timeresolved ICP-MS requires minimal sample preparation and provides immediate information on the metal content of individual cells and the related variance in a population of cells⁹ (see the Supporting Information). Intact cells are introduced directly into the ICP. ICP-MS spikes, corresponding to individual cells, are produced. The sensitivity of single-cell analysis using time-resolved ICP-MS is high: 10⁶ atoms/cell can be readily detected. The technique is also relatively robust with respect to the distribution of spike intensities. The spike intensity is proportional to the quantity of the analyte ions in the cell. The peak maximum of the spike intensity distribution can be used for the determination of the average metal content of the cells, as the peak maximum is relatively insensitive to the presence of sporadic large spikes in the distribution.¹⁰ Importantly, the average metal content in the cells can be quantitatively determined from the peak maximum using polydisperse metal nanoparticles as a calibration standard.⁹

In view of the medical value of Bi-based drugs toward the gastricdisease-causative agent *H. pylori*, their pharmaco-metallomic profiles were studied in this work using time-resolved ICP-MS. The application of such a methodology to track metallodrug uptake was demonstrated, and the initiative paves the way for building up a macroscopic view of the drug uptake process on a single-cell basis.

The feasibility of monitoring individual cells was demonstrated by tracking Mg^{II}, the most abundant divalent cation in living cells. For example, the number of Mg atoms in E. coli was previously measured to be on the order of 10^8 per cell.¹¹ Figure 1A shows the ICP-MS temporal profile of 24 Mg (abundance 78.99%)¹² for a suspension of wild-type *H. pylori* having a cell number density of 10⁵ cells/mL. Mg spikes, corresponding to individual H. pylori cells, were observed. The average frequency of the spikes was ~ 1 spike/s, or 0.01 spikes every 10 ms (the integration window of the ICP-MS instrument). Assuming a Poisson distribution of the appearance of spikes, the chance of two spikes overlapping in the ICP within the integration window was 0.005%. Overlap of ICP-MS intensity spikes within the ICP-MS integration window was thus insignificant. Each Mg spike should correspond to one H. pylori cell only. As shown in Figure 1E, the distribution of ²⁴Mg spike intensity of the *H. pylori* suspension was approximately log-normal with peak maximum of 2.04, which corresponds to 2.85×10^7 Mg atoms/cell using MgO nanoparticles as calibration standard as we did previously.9 To validate the methodology, the ²⁴Mg content was also determined using a batchtype method and found to be 2.52×10^7 Mg atoms/cell, in agreement with the single-cell measurement. In addition, the distribution showed a full width at half-maximum (fwhm) of 0.42, indicative of biological variations in a population of cells. ICP-MS intensity spikes of ²⁴Mg can therefore serve as a marker for the presence of intact *H. pylori* cells.

The uptake of metallodrug was exemplified using CBS, which is the mainstay for the eradication of *H. pylori*. The inhibition of *H. pylori* by the metallodrug has been associated with the disruption of the functions of key proteins.¹³ No ²⁰⁹Bi (abundance 100%)¹² ICP-MS

Received:February 11, 2011Published:April 25, 2011



Figure 1. Feasibility of using time-resolved ICP-MS for tracking the presence of individual cells and metallodrug uptake. (top) ICP-MS temporal profile of an *H. pylori* suspension. ²⁴Mg and ²⁰⁹Bi were measured from *H. pylori* grown in Brucella broth with (A, B) 0 and (C, D) 20 μ g/mL of CBS, respectively. (bottom) Corresponding distributions of ICP-MS spike intensities for ²⁴Mg and ²⁰⁹Bi.

intensity spikes were observed for the H. pylori suspension prior to the incubation of the metallodrug (Figure 1B), suggesting that no Bi is present in the pathogen under normal conditions. In contrast, ²⁰⁹Bi spikes corresponding to Bi in individual H. pylori cells were readily detected from H. pylori treated with 20 µg/mL CBS $(\sim 30 \,\mu\text{M Bi})$ for 16 h (Figure 1D). No ²⁰⁹Bi spikes were detected for a solution containing only 20 μ g/mL CBS.²⁴Mg spikes for the treated H. pylori were also measured separately. For the H. pylori suspension having a cell density of 10⁵ cells/mL, the frequency of the Bi spikes was \sim 1 spike/s, the same as the Mg spike frequency for untreated *H. pylori*. The distribution of ²⁰⁹Bi spike intensity for the H. pylori suspension was also approximately log-normal with a peak maximum of 1.74 and fwhm of 0.40 (Figure 1F). Interestingly, the Mg spike frequency was reduced to ~ 0.7 spike/s after CBS treatment (Figure 1C). The reduction was probably due to the loss of live and/or intact H. pylori cells. The Bi spikes in Figure 1D correspond to Bi in individual H. pylori cells. The fwhm is comparable to that of the Mg distribution.

To investigate the potential effect of ferric ions on the uptake of the bismuth drug as previously proposed,¹⁴ a competition study involving ferric citrate and CBS was performed, and the potential interference by citrate was also evaluated. *H. pylori* cells treated with 20 μ g/mL CBS in the presence of 0, 25, 50, and 100 μ M Fe^{III} (as ammonium ferric citrate) for 16 h were cultured and washed prior to the ICP-MS measurement. The peak maximum of the ²⁰⁹Bi spike intensity distribution (corresponding to the average Bi content in the cells) was found to decrease monotonically with increasing concentration of Fe^{III}, whereas the growth of *H. pylori* (as a percentage), measured in terms of the optical density at 600 nm (OD₆₀₀) of the *H. pylori* suspension, increased slightly (~10%) under the same conditions (Figure 2A). In contrast, almost no changes in the peak maximum



Figure 2. Protective effect of ferric citrate against CBS accumulation in *H. pylori*, as demonstrated by plots of the normalized peak maximum of the ²⁰⁹Bi spike intensity distribution (**■**) and the percentage growth (**●**) of *H. pylori* as functions of the concentrations of (A) Fe^{III} (as ammonium ferric citrate) and (B) citrate (as sodium citrate).



Figure 3. Kinetics of CBS uptake in *H. pylori* and the relationship between number of spikes and cell viability. (A) Normalized peak maximum of the ²⁰⁹Bi spike intensity distribution vs incubation time for *H. pylori* treated with 20 μ g/mL CBS. (B) Normalized peak maximum of ²⁰⁹Bi spike intensity distribution (**II**) and percentage growth (**●**) and (C) the number of ²⁰⁹Bi spikes measured in 60 s as functions of CBS concentration.

and percentage growth were observed for *H. pylori* suspensions treated with 20 μ g/mL CBS in the presence of different amounts of sodium citrate (Figure 2B), indicating that citrate was responsible for neither the growth nor the interference with the metallodrug uptake. Indeed, Fe^{III} protects *H. pylori* from Bi stress in a concentration-dependent fashion. Such protection may be attributed to competition in the transport of the metals into the cells through, for example, the ferric citrate outer membrane receptor Fec A, which is responsible for ferric citrate translocation.¹⁵

The time- and dose-related bismuth uptake upon CBS treatment in *H. pylori* was monitored to investigate the correlation between bismuth intake and cell viability. To study the kinetics of the metallodrug uptake, *H. pylori* cultured in the presence of 20 μ g/mL CBS was prepared, harvested, and washed after 0, 1, 2, 4, 6, 8, 16, and 24 h, respectively. The suspensions were subjected to time-resolved ICP-MS measurements of ²⁰⁹Bi immediately after washing. As shown in Figure 3A, the peak maximum of the ²⁰⁹Bi spike intensity increased gradually within the first 8 h of incubation and leveled off after 8 h. The uptake of Bi was relatively slow, with a half-life of ~3 h.

The dependence of CBS concentration on Bi uptake was subsequently examined in *H. pylori* suspensions after treatment with 0, 5, 10, 20, 40, and 80 μ g/mL CBS for 16 h. As shown in Figure 3B, the accumulation of Bi was linear for [CBS] < 20 μ g/mL and then gradually reached a maximum. The peak maximum of the ²⁰⁹Bi spike intensity distribution increased with [CBS] from 0–40 μ g/mL and became saturated for [CBS] > 40 μ g/mL. Since no appropriate bismuth nanoparticle was available for quantification, a batch-type method was used to determine the metal content. A digested sample of *H. pylori* treated with 20 μ g/mL CBS showed an accumulation of 9.77 × 10⁵ Bi atoms/cell, of which a significant fraction of Bi entered the periplasm and cytoplasm (see the Supporting Information).

Interestingly, the growth of *H. pylori* (as indicated by OD₆₀₀) was enhanced in the presence of low concentrations of CBS (<10 μ g/mL) and started to drop below 100% at higher drug concentrations. No further inhibition was observed for [CBS] > 40 μ g/mL. The same concentration range also revealed a constant peak maximum of the ²⁰⁹Bi spike intensity. Our results show that uptake of a small amount of Bi (i.e., <50% of Bi uptake limit) stimulates the growth of *H. pylori*, and the growth is inhibited afterward. The change in the number of ²⁰⁹Bi spikes as a function of the CBS concentration followed a similar profile as the percentage growth of *H. pylori* (Figure 3B,C). The number of Bi spikes reflects the population of *H. pylori* upon CBS treatment, which is consistent with the OD measurement since both methods monitor the total biomass in the samples.

In summary, we have demonstrated that the changes in bismuth content in bacteria with volumes in femtoliter range can be rapidly monitored and that single-cell analysis offers valuable biologically relevant insights into the uptake of bismuth-based drugs. The changes in the peak maximum of the spike intensity distribution with time provided a snapshot of the bismuth uptake in *H. pylori*. The large variation observed for both the ²⁴Mg and ²⁰⁹Bi spike intensities showed that cells were growing at different stages in life cycle. Importantly, the competition between ferric ion and bismuth was demonstrated, indicative of a competitive transport pathway involving the two metals in *H. pylori*. In addition, the study has launched new directions in using time-resolved ICP-MS for extensive single-cell-based applications, including in vivo simultaneous multielemental monitoring,¹⁶ rapid diagnosis of metal-relevant diseases, investigation of cellular heterogeneity,¹⁷ and identification of metal-related genes.

ASSOCIATED CONTENT

Supporting Information. Experimental details and bismuth content determination. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author hsun@hku.hk; wtchan@hku.hk

Author Contributions ⁺These authors contributed equally.

ACKNOWLEDGMENT

This work was supported by the Research Grants Council of Hong Kong SAR, P. R. China (Projects HKU7043/06P, HKU1/07C, HKU7042/07P, HKU7006/09P, HKU7049/09P, and $N_{\rm HKU752/09}$), the Croucher Foundation, and the Seed Funding Scheme for Basic Research of the University of Hong Kong. C.N.T. is grateful to F. Ng and Prof. B. J. Zheng from the LKS Faculty of Medicine, HKU, for assistance with *H. pylori* culture at the initial stage of the project.

REFERENCES

(1) Bertini, I.; Gray, H. B.; Stiefel, E. I.; Valentine, J. S. *Biological Inorganic Chemistry: Structure and Reactivity*; University Science Books: Sausalito, CA, 2007.

(2) (a) La Fontaine, S.; Ackland, M. L.; Mercer, J. F. B. *Int. J. Biochem. Cell Biol.* 2010, 42, 206. (b) Rouault, T. A. *Nat. Chem. Biol.* 2006, 2, 406. (c) Kozlowski, H.; Janicka-Klos, A.; Brasun, J.; Gaggelli, E.; Valensin, D.; Valensin, G. *Coord. Chem. Rev.* 2009, 253, 2665.

(3) (a) Guo, Z.; Sadler, P. J. Angew. Chem., Int. Ed. 1999, 38, 1512.
(b) Thompson, T. K.; Orvig, C. Science 2003, 300, 936. (c) Lippard, S. J. Nat. Chem. Biol. 2006, 2, 504.

(4) Rosenberg, B.; VanCamp, L.; Trosko, J. E.; Mansour, V. H. Nature 1969, 222, 385.

(5) (a) Yang, N.; Sun, H. *Coord. Chem. Rev.* **2007**, *251*, 2354. (b) Ge, R.; Sun, H. *Acc. Chem. Res.* **2007**, *40*, 267.

(6) (a) Shi, W.; Chance, M. R. Cell. Mol. Life Sci. 2008, 65, 3040.
(b) Salt, D. E.; Baxter, I.; Lahner, B. Annu. Rev. Plant Biol. 2008, 59, 709.
(c) Mounicou, S.; Szpunar, J.; Lobinski, R. Chem. Soc. Rev. 2009, 38, 1119. (d) Sun, X.; Tsang, C. N.; Sun, H. Metallomics 2009, 1, 25.(e) Chen, C.; Chai, Z.; Gao, Y. Nuclear Analytical Techniques for Metallomics and Metalloproteomics; RSC Publishing: Cambridge, U.K., 2010. (f) Kim, A. M.; Vogt, S.; O'Halloran, T. V.; Woodruff, T. K. Nat. Chem. Biol. 2010, 6, 674. (g) Sun, H.; Chai, Z. F. Annu. Rep. Prog. Chem., Sect. A: Inorg. Chem. 2010, 106, 20.

(7) Outten, C. E.; O'Halloran, T. V. Science 2001, 292, 2488.

(8) (a) Lahner, B.; Gong, J.; Mahmoudian, M.; Smith, E. L.; Abid, K. B.; Rogers, E. E.; Guerinot, M. L.; Harper, J. F.; Ward, J. M.; McIntyre, L.; Schroeder, J. I.; Salt, D. E. *Nat. Biotechnol.* **2003**, *21*, 1215. (b) Danku, J. M. C.; Gumaelius, L.; Baxter, I.; Salt, D. E. *J. Anal. At. Spectrom.* **2009**, *24*, 103.

(9) Ho, K. S.; Chan, W. T. J. Anal. At. Spectrom. 2010, 25, 1114.

(10) Rubin, A. Statistics for Evidence-Based Practice and Evaluation; Cengage Learning: Belmont, CA, 2009.

(11) Finney, L. A.; O'Halloran, T. V. Science 2003, 300, 931.

(12) CRC Handbook of Chemistry and Physics, 79th ed.; Lide, D. R., Ed.; CRC Press: Boca Raton, FL, 1998.

(13) (a) Zhang, L.; Mulrooney, S. B.; Leung, A. F. K.; Zeng, Y.; Ko, B. B. C.; Hausinger, R. P.; Sun, H. *Biometals* **2006**, *19*, 503. (b) Cun, S.; Sun, H. *Proc. Natl. Acad. Sci. U.S.A.* **2010**, *107*, 4943.

(14) (a) Husson, M. O.; Legarand, D.; Spik, G.; Leclerc, H. Infect. Immun. 1993, 61, 2694. (b) Domenico, P.; Reich, J.; Madonia, W.; Cunha, B. A. J. Antimicrob. Chemother. 1996, 38, 1031.

(15) (a) van Vliet, A. H. M.; Bereswill, S.; Kusters, J. G. In *Helicobacter pylori: Physiology and Genetics*; Mobley, H. L. T., Mendz, G. L., Hazell, S. L., Eds.; ASM Press: Washington, DC, 2001; p 193. (b) Wandersman, C.; Delepelaire, P. *Annu. Rev. Microbiol.* **2004**, *58*, 611.

(16) Solyom, D. A.; Burgoyne, T. W.; Hieftje, G. M. J. Anal. At. Spectrom. 1999, 14, 1101.

(17) Wang, D.; Bodovitz, S. Trends Biotechnol. 2010, 28, 281.